

Functional Graphene structures for Energy-conversion Devices

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We have developed a facile, reliable and reproducible electrochemical method for direct preparation of functional graphene quantum dots (GQDs), which have a uniform size of 3–5 nm and exhibit a green luminescence. The O-containing groups on the surface of GQDs makes them soluble in aqueous media, facilitating further functionalization and various applications. As a novel electron acceptor material, the as-prepared GQDs have been integrated into a P3HT based solar cell, significantly enhancing the device performance. We also prepared nitrogen-doped GQDs (N-GQDs) with oxygen-rich functional groups. Unlike their N-free counterparts, the newly produced N-GQDs with a N/C atomic ratio of ca. 4.3% emit blue luminescence and possess an electrocatalytic activity comparable to that of a commercially available Pt/C catalyst for the oxygen reduction reaction (ORR) in an alkaline medium. In addition to their use as metal-free ORR catalysts in fuel cells, the superior luminescence characteristic of N-GQDs allows them to be used for biomedical imaging and other optoelectronic applications.

A new complex catalyst system of ternary Pt/PdCu nanoboxes anchored onto 3D graphene sheets has also been fabricated by a dual solvothermal process. The electrocatalytic activity of the Pt/PdCu/3DGF for ethanol oxidation are not only significantly higher than that of pure Pt and PdCu electrodes, but also has an about 4-fold improvement over the well-established commercial Pt/C catalysts (E-TEK 20% Pt/C) as normalized to the total mass of active metals, which, in combination with the demonstrated single cell, shows the great potential of the geometry-defined Pt/PdCu/3DGF as excellent electrocatalysts for ethanol electrooxidation in alkaline media for direct ethanol fuel cells.

Further, a versatile, ultralight, N-doped, 3D graphene framework has been developed, which mainly consists of the network of only few graphene layers and has an ultra-low density of ca. 2.1 mg/cm³. Its adsorption capacity is as high as 200–600 times its own weight for common pollution and organic solvents, much higher than that of the best carbonaceous sorbents reported previously. Based on the synergetic function of 3D open-pore structure and N doping, the GF supercapacitor has generated a high specific capacitance of 484 F/g, far superior to the typical carbon-based electrodes. Besides the potential as a new metal-free catalyst for efficient electrocatalytic ORR demonstrated in this study, the ultralight GF also provides an important platform for developing a variety of advanced devices, such as sensors and batteries.

References

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