Core/Shell Co/Fe₃O₄ Nanoparticles as an Active and Durable Catalyst for the Oxygen Reduction Reaction in Alkaline Media <u>Chen-Hao Wang^{1*}</u>, Chih-Wei Yang¹, Sun-Tang Chang¹, Hsin-Chih Huang¹, Hsin-Cheng Hsu¹. ¹Department of Materials Science and Engineering, National Taiwan University of Science and Technology, Taipei, 10607, Taiwan

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Alkaline-based polymer electrolyte fuel cell (PEFC) has lower cost than acid-based PEFC¹. In this study, the core/shell electrocatalyst has been synthesized to show high efficiency and durability during oxygen reduction reaction (ORR) in alkaline media. Dodelet et al. used iron-based electrocatalysts at pyrolyzed temperature of 1050°C, They showed that their catalyst had good performance and high open circuit voltage $(OCV)^2$. On the other hand, Wu et al. utilized polyaniline as a precursor for a carbon-nitrogen template treated in hightemperature (1050 °C) pyrolysis and incorporated by iron and cobalt, which exhibited high activity and remarkable In this study, we develop a Core-shell stability³. structural catalyst process which offers a low temperature synthesis process and without the need of further pyrolysis step.

The resulting Co/Fe₃O₄ core/shell catalyst was characterized by X-ray photoelectron spectroscopy (XPS) and X-ray absorption (XAS) to confirm its core/shell structure. Rotating disk electrode (RDE) results reveals that the ORR catalyzed by Co/Fe₃O₄/C core/shell presented high ORR activity with current density of 5 mA cm⁻² at 0.2 V (vs. RHE, 0.1 M KOH) and electron transfer number of 3.95 which indicates a direct four-electron pathway for the reduction of O₂ to H₂O over the Co/Fe₃O₄ core/shell catalyst.

Reference

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Figure 1 The ORR curves of Co nanoparticle, Fe_3O_4 , and Co/Fe_3O_4 core/shell electrocatalyst in saturated oxygen, 0.1 M KOH. Scan rate: 10 mV/s; rotation speed: 1600 rpm.



Figure 2 Fourier transforms of k^3 -weighted EXAFS data at the (a) Fe and (b) Co K-edge for Co nanoparticle, Fe₃O₄, and Co/Fe₃O₄ core/shell electrocatalyst.