

Supramolecular Chemistry of Carbon Nanostructures: Concave-convex Interactions

Nazario Martín

Departamento de Química Orgánica I (UCM)
Ciudad Universitaria, s/n; 28040 Madrid- Spain,
IMDEA Nanociencia, Ciudad Universitaria de
Cantoblanco, 28049, Madrid- Spain.
www.ucm.es/info/fullerene/

e-mail: nazmar@quim.ucm.es

The readily available electron donor exTTF molecule has proved its efficiency for the design of unprecedented receptors for fullerenes and other carbon nanoforms. In this regard, custom-made tweezers and, particularly, macrocyclic receptors for fullerenes are proving a valuable alternative to achieve the affinity and selectivity required to meet goals such as the selective extraction of higher fullerenes, their chiral resolution or the self-assembly of functional molecular materials.

In this presentation some of the important breakthroughs based on electroactive TTF-type derivatives as supramolecular receptors for fullerenes and carbon nanotubes (CNTs) will be highlighted. Bowl and belt-shaped fullerene receptors based on this concave-convex complementarity principle will be presented. Other related and more sofisticated supramolecular assemblies formed by macrocycles endowed with exTTF concave geometry and convex fullerene surfaces will be discussed.[1] This will open the question if the concave-convex interactions really exists.

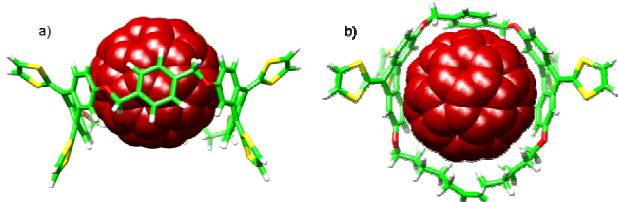


Figure. Side view (a) and top view (b) of the exTTFmacrocycle·C₆₀ associate.

The recognition motives have also been applied to carbon nanotubes with the aim of modifying their

References:

- [1] a) Pérez, E. M.; Sánchez, L.; Fernández, G.; Martín, N. *J. Am. Chem. Soc.*, **2006**, *128*, 7172. b) Pérez, E. M.; Martín, N. *Chem. Soc. Rev.*, **2008**, *37*, 1512. c) Fernández, G.; Pérez, E. M.; Sánchez, L.; Martín, N. *Angew. Chem., Int. Ed.*, **2008**, *47*, 1094. d) Isla, H.; Gallego, M.; Pérez, E. M.; Viruela, R.; Ortí, E.; Martín, N. *J. Am. Chem. Soc.*, **2010**, *132*, 1772. e) Huerta, E.; Isla, H.; Pérez, E. M.; Bo, C.; Martín, N.; de Mendoza, J. *J. Am. Chem. Soc.*, **2010**, *132*, 5351. f) Grimm, B.; Santos, J.; Illescas, B. M.; Muñoz, A.; Guldi, D. M.; Martín, N. *J. Am. Chem. Soc.*, **2010**, *132*, 17387. g) Canevet, D.; Gallego, M.; Isla, H.; de Juan, A.; Pérez, E. M.; Martín, N. *J. Am. Chem. Soc.*, **2011**, *133*, 3184. h) Isla, H.; Grimm, B.; Pérez, E. M.; Torres, M. R.; Herranz, M. A.; Viruela, R.; Aragó, J.; Ortí, E.; Guldi, D. M.; Martín, N. *Chem. Sci.*, **2012**, *3*, 498. i) For a recent review, see: Canevet, D.; Pérez, E. M.; Martín, N. *Angew. Chem. Int. Ed.* **2011**, *50*, 9248 – 9259.
- [2] a) Herranz, M. A.; Ehli, C.; Campidelli, S.; Gutiérrez, M.; Hug, G. L.; Ohkubo, K.; Fukuzumi, S.; Prato, M.; Martín, N.; Guldi, D. M. *J. Am. Chem. Soc.*, **2008**, *130*, 66. b) C. Romero-Nieto, R. García, M. A. Herranz, Ch. Ehli, M. Ruppert, A. Hirsch, D. M. Guldi, N. Martín, *J. Am. Chem. Soc.*, **2012**, Article ASAP, DOI: 10.1021/ja211362z.
- [3] a) López, J. L.; Atienza, C; Seitz, W.; Guldi, D. M.; Martín, N. *Angew. Chem. Int. Ed.* **2010**, *49*, 9876 – 9880. b) López, J. L.; Atienza, C.; Insuasty, A.; López-Andarias, J.; Romero-Nieto, C.; Guldi, D. M.; Martín, N. *Angew. Chem. Int. Ed.* **2012**, *51*, 3857 – 3861.

electronic properties,[2] as well as for the hierarchical organization of nano and mesoscopic 3D helical fibers.[3]

