

ALD and pulsed-CVD of Ru, RuO₂, and SrRuO₃[#]Jeong Hwan Han[§] and Cheol Seong Hwang^{*}

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As technology road map for DRAM capacitor with a design rule below 20 nm requires t_{ox} of less than 0.4 nm, noble metals, such as Ru, and higher-k dielectrics, TiO₂ and SrTiO₃, have gained a lot of interest as innovative materials for future DRAM. Among various electrode candidates, Ru, RuO₂, and SrRuO₃ have been spotlighted due to their low resistivity (Ru~7 $\mu\Omega\cdot\text{cm}$, RuO₂~30 $\mu\Omega\cdot\text{cm}$, SrRuO₃~300 $\mu\Omega\cdot\text{cm}$), high chemical stability, and high work function (Ru~4.7 eV, RuO₂~5.1 eV, SrRuO₃~5.1 eV). Atomic layer deposition (ALD) and chemical vapor deposition (CVD) are techniques of choice for fabricating nano-scale Ru based films because of their excellent thickness controllability and conformal growth characteristic over the three-dimensional substrate.

In this presentation, the authors will report on the growth behaviors and characteristics of Ru film by ALD and pulsed-CVD using MO precursor, Ru(DMPD)(EtCp), and inorganic RuO₄ precursor, respectively. Pulse-CVD Ru film showed enhanced mass-production compatibility, where a higher growth rate of 0.18 nm/cycle was obtained than that of ALD Ru. Additionally, growth of RuO₂ and SrRuO₃ electrodes were explored by pulsed-CVD and combined ALD/CVD process using RuO₄ and Sr(iPr₃Cp)₂ as Ru and Sr precursor, respectively. Pulsed-CVD RuO₂ film showed growth rate of 0.24 nm/cycle and excellent conformality over a hole structured substrate with a opening diameter of 100 nm and aspect ratio of 10 as shown in Fig. 1 (a). [1] Study on the initial growth behavior of pulsed-CVD RuO₂ revealed that highly oxidative and electronegative properties of RuO₄ precursor results in rather different nucleation behavior from that of ALD Ru(RuO₂) using MO precursor. Electrical properties of (Al-doped) TiO₂ films grown on Ru and RuO₂ are compared. Figure 1 (b) shows that Al-doped TiO₂ film has lower leakage current density of on RuO₂ bottom electrode compared to that on Ru electrode even with slightly lower t_{ox} value. [2]

Stoichiometric SrRuO₃ film was deposited by alternating ALD SrO and pulsed-CVD RuO₂ at 230 °C. The cation ratio of Sr/Ru was controlled by modulating the SrO/RuO₂ sub-cycle ratio. To obtain crystallized SrRuO₃ film, post deposition annealing at 600-700 °C was performed. Figure 2 (a) and (b) shows GAXRD patterns and HRTEM images of Pt/SrTiO₃/SrRuO₃ capacitor, respectively. In-situ crystallization of ALD SrTiO₃ was achieved on crystallized SrRuO₃ film, indicating crystallized SrRuO₃ electrode plays a role as a seed layer for perovskite SrTiO₃ growth. [3]

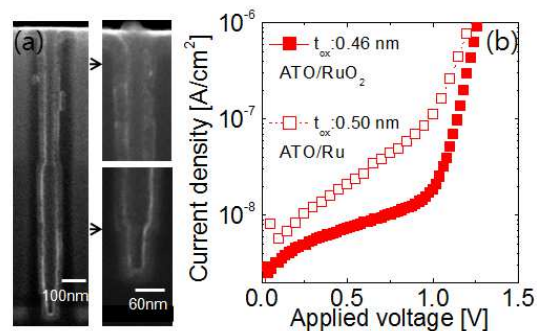


Figure 1 (a) The SEM images showing film conformality of pulsed-CVD RuO₂ over a hole structured substrate with an opening diameter of 100 nm and aspect ratio of 10. (Data reproduced from Ref. 1) (b) Leakage current density of ATO on Ru and RuO₂ bottom electrodes. (Data reproduced from Ref. 2)

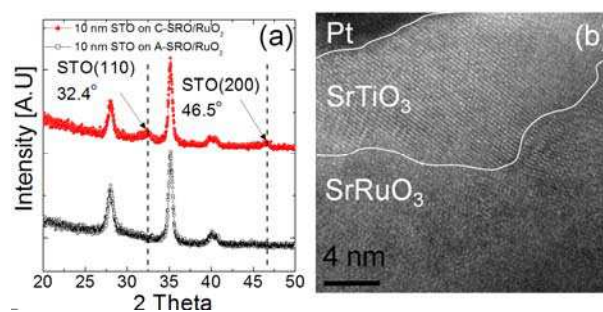


Figure 2 (a) GAXRD of 10-nm-thick ALD SrTiO₃ films grown on crystallized SrRuO₃ and amorphous SrRuO₃ electrodes. (b) HRTEM images of Pt/SrTiO₃/SrRuO₃ capacitor, showing the STO is in-situ crystallized. (Data reproduced from Ref. 3)

References

- [1] J. H. Han et al., *Chem. Mater.*, 22, 5700, (2010)
- [2] J. H. Han et al., *Appl. Phys. Lett.*, 99, 022901, (2011)
- [3] J. H. Han et al., *Chem. Mater.*, 24, 4686, (2012)

[#]Invited presentation

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