

Enhanced Supercapacitor Performance of Graphene/V₂O₅ Nanocomposites

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With increasing energy and power demands in high power applications such as, electric vehicles, hybrid electric vehicles and mobile electronics have stimulated significant research efforts on the development of new electrode materials for advanced energy storage devices [1]. Among them supercapacitor is an important energy storage device mainly due to their high power densities, fast recharge capability and long cycle life than secondary batteries [2]. Nowadays many researchers have been focused on the supercapacitor to increase their energy densities as well as decrease the fabrication costs. The electrode materials play an important role in the development of high performance supercapacitors in terms of the morphology, size, porosity and so forth. Among various electrode materials, graphene (electrochemical double layer) and vanadium pentoxide (V₂O₅, pseudocapacitors) are considered to be promising electrode materials for supercapacitor applications due to their intriguing characteristics such as low cost, and natural abundance [3]. Graphene exhibited high surface area, electrical conductivity and extraordinary thermal, mechanical, and electrical properties [4]. On the other hand V₂O₅ has layered structure, modest electronic conductivity and the satisfactory charge storage capacity [5]. By combing the attractive features of the graphene and V₂O₅ materials to enhance the electronic conductivity of V₂O₅ and maintain high electrolyte penetration/diffusion rates which leads to the improved capacitance.

In the present study, graphene/V₂O₅ nanocomposites were synthesized by a facile, fast and scalable microwave-assisted method for electrochemical supercapacitor applications. The structural, composition and morphological properties of the as prepared graphene/V₂O₅ nanocomposites were characterized by X-ray diffraction (XRD), laser Raman microscopy, X-ray photoelectron spectroscopy (XPS), field emission scanning electron microscopy (FE-SEM), and transmission electron microscopy (TEM). The electrochemical properties were examined by cyclic voltammetry (CV), galvanostatic charge- discharge and electrochemical impedance spectroscopy (EIS) measurements in 1M Na₂SO₄ aqueous solution.

Fig.1. shows the FE-SEM image of the as prepared graphene/V₂O₅ nanocomposites. It can be observed that the V₂O₅ nanoparticles are uniformly distributed on the surface of graphene nanosheets. The size of the nanoparticles was in the range of 20-30 nm. CV curves (**Fig.2.**) of V₂O₅, reduced graphene oxide and graphene/V₂O₅ nanocomposite electrodes exhibited similar and ideally rectangular shape, confirming an ideal capacitive behavior of the electrode. In addition, it can be

seen that the graphene/V₂O₅ nanocomposites electrode show higher integrated area than other electrodes, which indicates that the graphene/V₂O₅ nanocomposites electrode shows an excellent electrochemical performance.

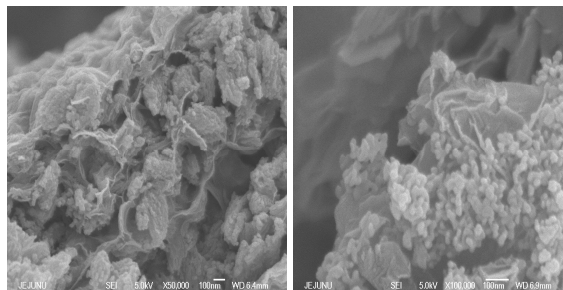


Fig.1. FE-SEM images of graphene/V₂O₅ nanocomposites.

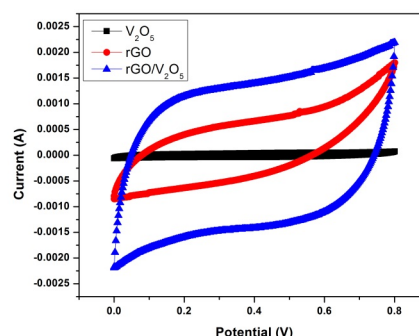


Fig.2. CV curves of V₂O₅, reduced graphene oxide and graphene/V₂O₅ nanocomposites at a scan rate of 5 mV s⁻¹.

Detailed results and discussion will be presented in the meeting.

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