

Development of Ultra-Low Loading Pt Alloy Cathode Catalyst for PEM Fuel Cells

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The main strategies to decrease the platinum loading of cathode electrodes are based on optimization of the electrode structures and implementation of more active Pt alloy catalyst and supports. Our strategy to decrease the platinum loading in the cathode is based on (i) synergetic effects of CCC (carbon composite catalyst, free of platinum) on which Pt or Pt alloy catalysts is deposited, thus forming two types of hybrid catalysts Pt/CCC and Pt-alloy/CCC (HCC) catalysts [1-5]. The goal of our second strategy is the development of activated graphitic carbon supported Pt alloy catalyst Pt-alloy/AGC (USC) or Pt-alloy/CNC -YU. The highly corrosion resistant graphitic carbon is developed at USC and at Yonsei University [5]. The goals are: (i) to achieve kinetic mass activity in H₂/O₂ fuel cell higher than the DOE 2017 target of 0.44 A mg_{PGM}⁻¹, (ii) to demonstrate durability of the kinetic activity per DOE cycling protocol between 0.6 and 1.0 V, and with performance loss less than 40% (iii) to accomplish high current density performance and durability in H₂/air fuel cell to meet 2017 DOE targets. Cell diagnostics of the cathode catalysts such as ECSA, ORR mass activity, and oxygen polarization curves were performed after 5,000, 10,000, 20,000, and 30,000 cycles. The support durability was studied by applying 1.2 V vs. NHE constant potential with respect to the anode for 400 hours according to the DOE protocol. Cell diagnostics of the cathode catalyst such as ECSA, ORR mass activity, and oxygen polarization curves were performed after 100, 200, 300 and 400 hours.

Pt catalyst was synthesized at USC which shows synergetic effect between the catalytically active non-platinum CCC support and Pt catalyst. CCC was synthesized with onset potential for oxygen reduction of 0.92 V vs. SHE and with less than 2.5% H₂O₂ production. Our Pt/CCC₁₄₀ exhibits mass activity of 0.29A/mg_{Pt} compared with 0.16 mass activity measured for 46%Pt/C (TKK) catalyst [6]. Mass activity loss after 200 h potential holding was only 24% compared with 58% loss observed for commercial catalyst after 24 hours. The potential holding experiment at 1.2 V for the support durability of Pt/CCC₁₄₀ showed only 6 mV loss after 200 hours indicating very stable support and catalyst. The observed voltage loss for commercial catalyst was higher than 250 mV after 200 h potential holding test at 1.2 V vs. SHE.

USC hollow graphitic carbon supported Pt catalyst, Pt/AGC after 400 h holding at 1.2 V vs. SHE shows only 33% loss of mass activity compared with 40% loss of 2017 DOE target. The platinum was deposited on in-house developed support synthesized using three-step procedure. The durability of high current performance of this catalyst under hydrogen air after 400 h potential holding at 1.2 V shows only 3mV loss at 800 mA compared with 30 mV loss of 2017 DOE target.

One dimensional simplified agglomerate-based macro-homogeneous model is developed and was solved analytically and numerically to explain the effect of the parameters that control the mass activity of the catalyst [7-8]. Table 2 summarizes the status of the low Pt alloy catalyst development research.

Table 2 Status of the low-Pt alloy catalyst development research (HCC₂₅₀ and Pt₂Ni₁/CNC catalysts)

Characteristics	2017 Targets	Status
PGM total loading	0.125 mg _{Pt} /cm ²	0.2 mg _{Pt} /cm ²
Mass activity (MA) 80°C, 100% RH, 150kPa _{abs.}	0.44A/mg @0.9V _{iR-free}	0.38A/mg _{Pt} <u>HCC₂₅₀</u> 0.44A/mg _{Pt} <u>YU Pt₂Ni₁/CNC</u>
Catalyst durability 30,000 cycles, 0.6-1.0 V, 50mV/s, H ₂ /N ₂ , 80°C, 100kPa _{abs.}	MA loss ≤40% ECSA loss ≤40% Current Density @0.8A/cm ² in H ₂ -air loss ≤ 30 mV	<u>USC HCC₂₅₀</u> *47% loss (MA) *15% loss (ECSA) *53 mV loss (H ₂ -air) <u>YU Pt₂Ni₁/CNC</u> *32% loss (MA) *26% loss (ECSA)
Support durability 1.2 V for 400h, 80°C, H ₂ -N ₂ , 150kPa _{abs.} , 100% RH	MA loss ≤40% ECSA loss ≤40% Current Density @0.8A/cm ² in H ₂ -air loss ≤ 30 mV	<u>USC HCC₂₅₀</u> *47% loss (MA) *42% loss (ECSA) *29 mV loss (H ₂ -air) <u>YU Pt₂Ni₁/CNC</u> *48% loss (MA) *43% loss (ECSA)

The current status of the research indicates that the results obtained in the first phase of the project demonstrate promising performance of the mass activity under H₂/O₂ and high current performance under H₂/air and durability of the kinetic mass activity and support stability. In the second phase two catalysts will be selected for further studies.

References

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