

## Formation of a bilayer of low-temperature CVD-SiO<sub>2</sub> and sputtered Al<sub>2</sub>O<sub>3</sub> films on polyethylene terephthalate substrates for an OLED encapsulation layer

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Organic light-emitting diode (OLED) displays have promising potential for flat-panel display to replace liquid crystal displays (LCDs). The limited lifetime of OLED is a major drawback for its commercial applications such as TV and monitor. Since the organic materials and low work-function metals used in OLEDs are sensitive to moisture, OLED displays are easily degraded in humid environment. Encapsulation with organic or inorganic thin films has been actively investigated to fulfill a requirement of low water permeability ( $< 10^{-6}$  g/m<sup>2</sup>·day). Recently, multilayers of inorganic thin films have been proposed for the encapsulation of OLEDs using various deposition methods such as sputtering, atomic layer deposition (ALD), and chemical vapor deposition (CVD).[1-5].

In this study, we prepared a single-layer barrier of sputtered Al<sub>2</sub>O<sub>3</sub> and a bilayer barrier consisting of CVD SiO<sub>2</sub> and sputtered Al<sub>2</sub>O<sub>3</sub> as a potential moisture barrier for an OLED encapsulation. Sputtered Al<sub>2</sub>O<sub>3</sub> films were prepared at room temperature using an RF magnetron sputtering system (CS5000, SNTek Co.). The process pressure and Ar flow rate were 5 mTorr and 10 sccm, respectively. The RF power of Al<sub>2</sub>O<sub>3</sub> target was 150 W. For the bilayer barrier, the CVD SiO<sub>2</sub> thin films were formed at 100°C using tris(ethyl-methyl-amino)silane (TEMS, HSi[N(C<sub>2</sub>H<sub>5</sub>)(CH<sub>3</sub>)<sub>3</sub>]<sub>3</sub>) and O<sub>3</sub> as a Si precursor and oxidant gas, respectively.

Figure 1 shows the Fourier transform infrared spectroscopy (FTIR) spectra of CVD SiO<sub>2</sub> layer. Absorbance bands at 1070 cm<sup>-1</sup> and 800 cm<sup>-1</sup> are ascribed to the stretching and bending vibrations of Si-O-Si, respectively. The peak at 950 cm<sup>-1</sup> has been reported to be associated with the stretching mode of Si-OH. The absorbance band at 880 cm<sup>-1</sup> is assigned to the bending modes of H-SiO<sub>3</sub>, indicating that CVD SiO<sub>2</sub> contains small amount of -OH radicals. Since CVD SiO<sub>2</sub> with -OH radicals exhibits a flowable behavior, the CVD-SiO<sub>2</sub> film exhibits a conformal gap-filling at the acute angle region between the silica particle and the substrate, as shown in Fig. 2. Because the conformal gap-filling behavior of CVD SiO<sub>2</sub> enhanced the blocking of the pin holes or cracks of sputtered Al<sub>2</sub>O<sub>3</sub> layer, the bilayer barrier with CVD-SiO<sub>2</sub> layer and sputtered Al<sub>2</sub>O<sub>3</sub> layer reduced the water permeation compared to the Al<sub>2</sub>O<sub>3</sub> single-layer barrier, as shown in Fig. 3. The bilayer with the 100 nm thick Al<sub>2</sub>O<sub>3</sub> and the 100 nm thick CVD-SiO<sub>2</sub> exhibits lower water vapor transmission rate (WVTR) (0.214 g/m<sup>2</sup>·day) than the 200 nm thick sputtered Al<sub>2</sub>O<sub>3</sub> single-layer (0.325 g/m<sup>2</sup>·day), although both barriers have the same thickness.

In summary, the encapsulation bilayer with CVD SiO<sub>2</sub> and sputtered Al<sub>2</sub>O<sub>3</sub> layers was studied to reduce the moisture permeation through PET film. Even though the WVTR (0.214 g/m<sup>2</sup>·day) of the bilayer barrier is much

lower than the target value (10<sup>-6</sup> g/m<sup>2</sup>·day) of the OLEDs, the inorganic bilayer barrier of OLEDs with different materials and different deposition methods makes it possible to reduce the water permeation to OLEDs. Especially the gap-filling property of flowable CVD-SiO<sub>2</sub> plays the key role in a surface planarization and blocking of surface defects.

### References

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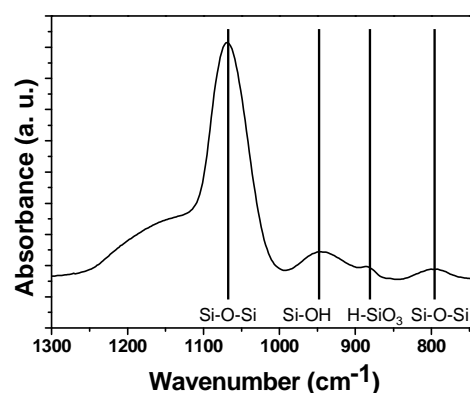


Fig. 1. The FTIR spectra of CVD-SiO<sub>2</sub> film.

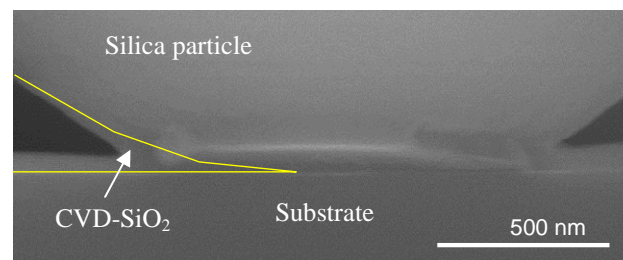


Fig. 2. Cross-sectional SEM image showing the coverage of CVD-SiO<sub>2</sub> film at the acute angle underneath a silica particle.

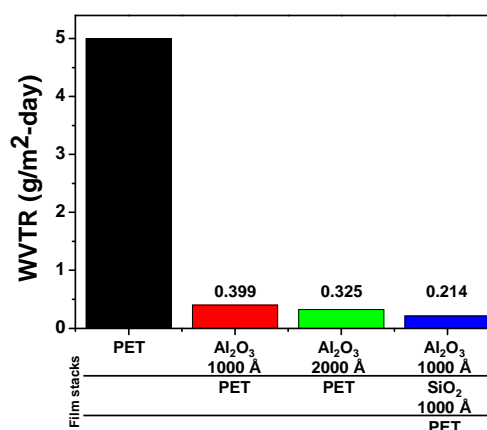


Fig. 3. The comparison of the WVTR of the single-layer and bilayer barriers.