

Light-accelerated TiO₂/Pt/Au Nanomotor Fabricated by Using Porous Aluminum Template

T.Sagisaka^a, Y. Hoshi^a, I. Shitanda^{a, b}, M. Itagaki^{a, b}

^a.Department of Pure and Applied Chemistry, Faculty of Science and Technology, Tokyo University of Science, 2641, Yamazaki, Noda, Chiba 278-8510, Japan.

^b.Research Institute for Science and Technology, Tokyo University of Science, 2641 Yamazaki, Noda, Chiba 278-8510, Japan.

Nanomotor is nanoscale devices propelled by the electrocatalytic decomposition of a chemical fuel. Such artificial nanomachines are currently the subject of intense interest due to their potential applications in nanomachinery, nanomedicine, nanoscale transport. Previously, a nanomotor was reported based on Au/Pt nanorod and propelled by electrocatalytic decomposition of the hydrogen peroxide fuel¹⁾ This mechanism suggests that the hydrogen peroxide reduction the anodic (catalytic) reaction on the platinum segment involves electron produces protons. And cathodic reaction on the gold segment consumes protons. Thus, electron and protons flux within, surrounding the rod toward the gold cathode (Figure 1). Because of strong affinity between protons and water, water flux is generated surrounding the nanorod from the platinum end to the gold end. So, nanomotor can propels toward platinum end. Such mechanism suggests good correlation between the mixed potential difference and the speed of bimetallic nanorod.

Motion control of nanomotor is essential for performing various tasks and diverse applications. In the present study, we newly developed a light-accelerated nanomotor based on TiO₂/Pt/Au nanorod. The nanomotor is prepared by template-directed electrodeposition within the cylindrical nanopores of a porous aluminum membrane followed by the template removal¹⁾.

Ag is plated within crlindrinal nanopores of a porous aluminum membrane. After that Au, Pt, TiO₂ are plated in this order²⁾. Then, the nanomotors in a porous membrane are annealed for 1 hour. Finally, TiO₂/Pt/Au nanomotors are obtained by dissolved a porous membrane in 1 M NaOH. Movement of TiO₂/Pt/Au nanomotors was conducted in 2% H₂O₂ solution under ultraviolet (UV) light irradiation.

Figure 2 shows the average speeds of the TiO₂/Pt/Au nanomotor with and without UV light irradiation. The speed of TiO₂/Pt/Au nanomotor became faster by UV light irradiation. One mechanism of Light acceleration mechanism is suggested that generation of the hydrogen peroxide fuel on the TiO₂ surface. The other mechanism is suggested that generation of radicals on the nanomotor's surface. The radicals generated were removed an impurity absorbed on the nanomotor surface. The relationship between length of TiO₂ rod and speed of the TiO₂/Pt/Au nanomotor will be shown in the conference.

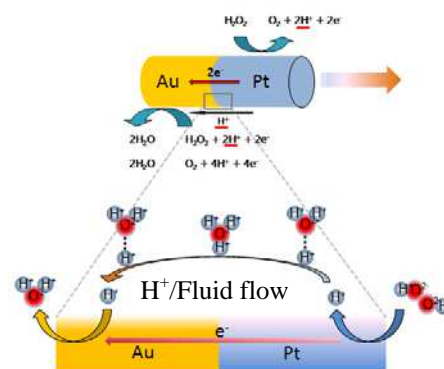


Figure 1 Driving Principle of Au/Pt nanomotor

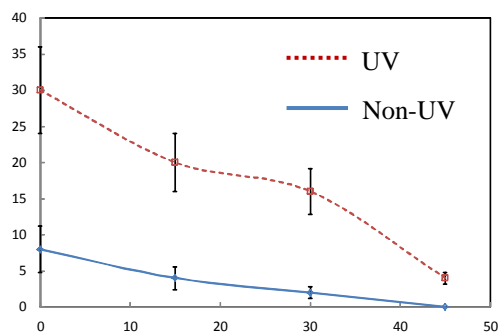


Figure 2 Speed of TiO₂/Pt/Au nanomotor with and without UV light irradiation

References

- 1) J. Wang, *ACS Nano*, **3** (2009).4
- 2) S. Karuppuchamy, et al, *Solid State Ionics*, **151** (2002).