

Development of Hermetic Sealant Material Based on the Temperature Distribution in IT-SOFC

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In this study, firstly, all heat source and amounts has been determined for electrolyte supported SOFC having 81 cm² active area and thus temperature distributions due to each heat source can computed and total temperature distribution clearly has been shown. For this, electrochemical, ohmic, activation and concentration polarization heat sources have been computed by using the boundary conditions taken from the experimental data such as current-voltage-power and mol fraction. Secondly, to provide hermetic seal SiO₂, CaO, Al₂O₃ and Na₂O based glass-ceramics has been produced and glass transition, softening temperatures and TEC have been tested by dilatometer and thus the most suitable sealant content has been selected.

There are some studies [1-9] on temperature distribution and heat sources, however, they has been investigate partial geometers such as one channel, one dimensions or some of the heat sources. There is no study about the effect of overall heat sources and temperature distributions on the sealant selection/production. The overall heat sources amount and distribution on the 9x9 cm² cell can be shown in Figure 1.

According temperature distribution, sealant material content was selected and molten at 1500 °C. Molten glass-ceramic and mold, produced from the furnace refractor material, can be shown in Figure 2.

[1] Martin Andersson, Hedvig Paradis, Jinliang Yuan, Bengt Sundén, Modeling Analysis of Different Renewable Fuels in an Anode Supported SOFC, Journal of Fuel Cell Science and Technology JUNE 2011, Vol. 8 / 031013-9

[2] Meng Ni, 2D thermal modeling of a solid oxide electrolyzer cell (SOEC) for syngas production by H₂O/CO₂ co-electrolysis, international journal of hydrogen energy 37 (2012) 6389e6399

[3] Daun K.J., Beale S.B., Liu F., Smallwood G.J. Radiation heat transfer in planar SOFC electrolytes. J. Power Sources 2006; 157: 302–10.

[4] COMSOL Multiphysics 3.5 user guide. Stockholm, Sweden; 2008.

[5] Jun LI, Ying-wei KANG, Guang-yi CAO, Xin-jian ZHU, Heng-yong TU, Jian LI, Numerical simulation of a direct internal reforming solid oxide fuel cell using computational fluid dynamics

method, Li et al. / J Zhejiang Univ Sci A 2008 9(7):961-969

[6] Anchasa Pramuanjaroenkij, Sadik Kakac, Xiang Yang Zhou, Mathematical analysis of planar solid oxide fuel cells, INTERNATIONAL JOURNAL OF HYDROGEN ENERGY 33 (2008) 2547– 2565.

[7] [1] Lu Y., Schaefer L., Li P. Numerical study of a flat-tube high power density solid oxide fuel cell Part I. Heat/mass transfer and fluid flow. J Power Sources 2005; 140: 331-9

[8] Wang Q., Li L., Wang C. Numerical Study of Thermoelectric Characteristics of a Planar Solid Oxide Fuel Cell with Direct Internal Reforming of Methane. J. Power Sources 2009; 186: 399-07

[9] Ferguson J., Fiard J., Herbin R. Three-dimensional numerical simulation for various geometries of solid oxide fuel cells. J Power Sources 1996; 58: 109-22.

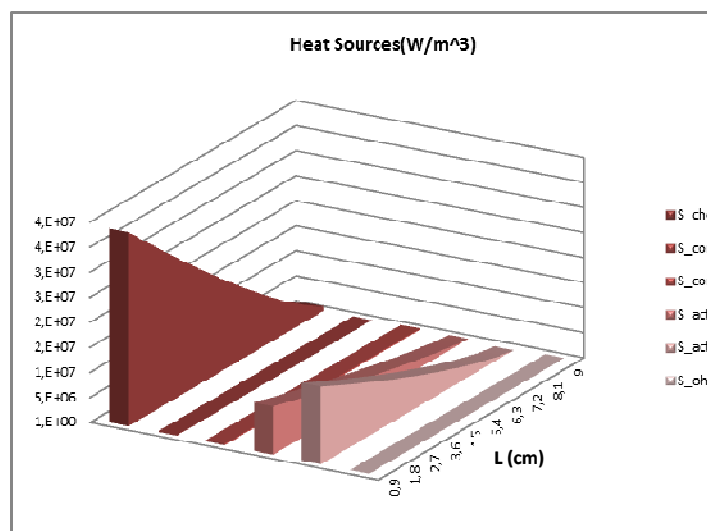


Figure 1. Different heat sources and distributions on the 81 cm² SOFC active area



Figure 2. Molten glass-ceramic (transparent appearance) in platinum crucible and mold for the sealant test samples