

Electrochemical Behaviors of SiO<sub>x</sub> Nanoparticles as an Anode Material for Lithium-ion Battery

Eunjun Park<sup>1</sup>, Min-Sik Park<sup>2</sup>, Young-Jun Kim<sup>2</sup>, Hansu Kim<sup>1,\*</sup>

<sup>1</sup> Department of Energy Engineering, Hanyang University,

<sup>2</sup> Advanced Batteries Research Center, Korea Electronics Technology Institute

<sup>1</sup> 222 Wangsimni-ro, Seongdong-gu, Seoul 133-791, Republic of Korea,

<sup>2</sup> 25, Saenari-ro, Bundang-gu, Seongnam-si, Gyeonggi-do, 463-816, Republic of Korea

Nano-sized SiO<sub>x</sub> nanoparticles were investigated as high capacity anode materials for lithium-ion batteries. High resolution transmission electron microscopy with x-ray diffraction analysis revealed that Si nanocrystals with the size of about 5 nm were well dispersed in amorphous SiO<sub>x</sub> matrix after heat treatment at 1200 °C under inert atmosphere. The electrochemical performances of these materials showed a reversible capacity of 950 mAh/g with stable capacity retention during 50 cycles.

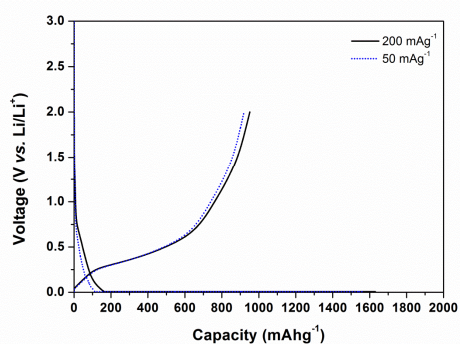


Fig.1. Charge and discharge curves of electrodes of the carbon-coated SiO<sub>x</sub> nanoparticles in the first cycle.