

Mixed ion and electron conducting ceramics for gas sensors

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Mixed ionic and electronic conducting perovskite-type $\text{BaM}_{0.33}\text{Nb}_{0.34-x}\text{Fe}_x\text{O}_{3-\delta}$ ($M = \text{Ca}, \text{Mg}$) have been prepared using solid state synthesis method. In-situ and ex-situ powder X-ray diffraction (PXRD) measurements at 30–800 °C showed high chemical stability of the synthesized materials under the CO_2 environment. Room temperature PXRD with Rietveld refinement and HR-TEM electron diffraction results confirmed the formation of perovskite-type $\text{BaMg}_{0.33}\text{Nb}_{0.34}\text{Fe}_x\text{O}_{3-\delta}$ ($Pm-3m$) and $\text{BaCa}_{0.33}\text{Nb}_{0.34-x}\text{Fe}_x\text{O}_{3-\delta}$ ($Fm-3m$) without any impurity phases. The sensing behavior has been characterized by electrochemical AC impedance spectroscopy and dc measurements at 300–700 °C by introducing different ppm level of CO_2 (0–1500 ppm) mixed with dry synthetic air. The increase in CO_2 level (ppm) lowered the resistance of the pellets. Interestingly, Fe-doping in $\text{BaM}_{0.33}\text{Nb}_{0.64}\text{O}_3$ improved the sensing properties by raising the total conductivity significantly (Fig. 1) [1, 2]. t_{90} , time required by sensor to reach 90% of its final stable reading (current density), also reduced by increasing the Fe-content. The highest t_{90} of 4 min for $\text{BaM}_{0.33}\text{Nb}_{0.33-x}\text{Fe}_x\text{O}_{3-\delta}$ ($x = 0.33$) was obtained under CO_2 atmosphere at 700 °C. $\text{BaCa}_{0.33}\text{Nb}_{0.33-x}\text{Fe}_x\text{O}_{3-\delta}$ ($x = 0.33$) exhibited an excellent long-term stability reassuring its possibility as a promising solid state CO_2 sensor.

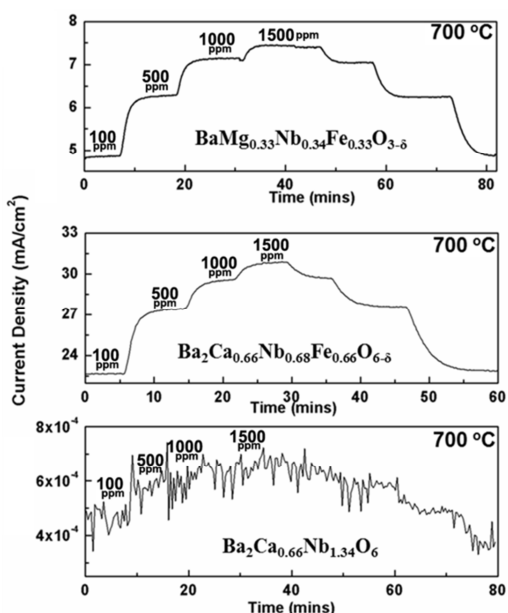


Figure 1. Response and recovery transients of $\text{BaCa}_{0.33}\text{Nb}_{0.67}\text{O}_3$, $\text{BaCa}_{0.33}\text{Nb}_{0.34-x}\text{Fe}_x\text{O}_{3-\delta}$ and $\text{BaMg}_{0.33}\text{Nb}_{0.34-x}\text{Fe}_x\text{O}_{3-\delta}$ at 700 °C using 3000 ppm CO_2 mixed in dry synthetic air [1, 2] (applied voltage = 0.1V).

References

1. R. Kannan, S. Mulmi, and V. Thangadurai, *J. Mater. Chem.*, DOI: 10.1039/C3TA10572E.
2. S. Mulmi and V. Thangadurai, *J. Electrochem. Soc.* 160, B95 (2013).