INFLUENCE OF THE AXIAL LIGAND ON IRON PHTHALOCYANINE TOWARD THE REDUCTION OF MOLECULAR OXYGEN.

<u>Ingrid Ponce</u>, Ruben Oñate, J. Francisco Silva, Carmen Castro, Marcos Caroli, José H. Zagal, Jorge Pavez.

Facultad de Química y Biología, Universidad de Santiago de Chile. Avenida Libertador Bernardo Ohiggins 3363, Estación Central, Santiago, Chile. e-mail: ingrid.ponce@usach.cl

Self assembled monolayers (SAMs) of 4 pyridinium salts functionalized with iron phthalocyanines complexes adsorbed on gold (111) electrodes have been constructed and characterized. Their activity for the reduction of molecular oxygen in aqueous solution has been examined. Electrochemical and scanning tunneling miscroscopy (STM) studies confirm the functionalization of the SAMs by iron phtalocianine complex. The electrocatalytic studies carries out on the Au(111)/SAMs/FePc electrode, showed that the O₂ reduction has shift to low potentials of almost 0.1V by the mere presence of the pyridiniums anchor, compared to Au(111)/FePc system. This effect is surprising since the presence of a spacer should decrese the overall electron transfer rate. Further, the structure of each molecule influence the electrocatalytic activity for the reduction of molecular oxygen.

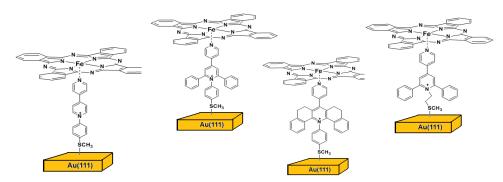


Fig. 1. SAMs of Pyridinium functionalized with Iron phthalocyanine, Au(111)/SAMs/FePc systems.

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References:

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