

Advanced MnO₂/CNT ultracapacitors- Transition from planar to micropatterned array electrodes

S. Raina¹, S. H. Hsu¹, S. Akbulut¹, M. Yilmaz¹, W. P. Kang¹, J. H. Huang²

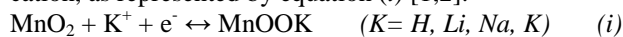
¹Dept. of Electrical Engineering and Computer Science, Vanderbilt University, Nashville, TN 37212, USA

email: supiraina@vanderbilt.edu

²Dept. of Materials Science and Engineering, National Tsing Hua University, Taiwan

Ultracapacitors with substantially higher power and energy densities, faster charge/discharge capability and longer cycle lifetime have received considerable attention [1,2]. CNTs excel as EDLCs, because they are excellent electrical conductors, have very large specific surface area to volume ratio, chemically and thermally stable, and can be fabricated at lower costs than other materials [1]. However, pure CNTs have low specific capacitance, usually below 40 F/g, depending on the purity, microstructure, and type of electrolyte [3].

Pseudocapacitive transition metal oxides, which produce higher capacitance than double layer carbonaceous materials, are being studied; among which conductive RuO₂ shows outstanding performance. However, its high cost hinders it from large-scale application. Research efforts have been focused on alternative low cost transition-metal oxide MnO₂ because of its high energy density, environmental compatibility and natural abundance [1-3]. The high theoretical specific capacitance is due to pseudocapacitive behavior involving rapid, reversible faradaic reactions where the oxidation state of Mn varies between +3 and +4 in conjunction with the intercalation and deintercalation of the electrolyte cation, as represented by equation (i) [1,2].



However, MnO₂ shows low capacitance without conductive additives due to its intrinsically poor electrical conductivity [2,3]. One possible solution involves the use of MnO₂ with multi-walled CNTs.

In this paper, we present fabrication and characterization of *planar and advanced micropatterned CNT/MnO₂ microelectrode array (MEA) electrodes* by direct electrochemical deposition of MnO₂ on vertically aligned CNTs. Such an array structure provides a 3-D surface allowing for better MnO₂ incorporation as well as enhanced mobility of the electrolyte to access the interior of the porous CNT/MnO₂ microelectrodes.

A highly doped n-type silicon substrate was first oxidized to form a thin-layer of SiO₂ followed by UV photolithography step to define the array layout. The exposed SiO₂ regions were etched using BOE solution and then thin-films of the buffer layer (Ti) and the catalyst (Co) were sputtered. After photo-resist lift-off, vertically aligned multi-walled CNTs were grown using hot-filament CVD process with methane as the carbon gas source. The height of the CNTs was controlled by varying the duration of CH₄ gas flow. This process can be adapted to fabricate arrays with different geometries and dimensions. The SEM images in figures 1(a) and (b) shows the planar MnO₂ coated CNTs and the porous MnO₂ microstructure. Before and after MnO₂ deposition SEM micrographs for the microelectrode array can be seen in figures 2(a) and (b), respectively.

Electrochemical characterization of the novel ultracapacitors was performed in a flat cell using cyclic voltammetry in a 3 electrode configuration with 0.1M KCl as the supporting electrolyte. Platinum coil was used as the counter electrode with a Ag/AgCl reference electrode. The cyclic voltammograms recorded at

50mV/s, before and after MnO₂ deposition, can be seen for the planar and the MEA electrodes in figures 1(c) and 2(c) respectively. The **planar electrode** delivers a maximum capacitance of 50mF/cm² or 30F/cm³ which represents a **10-fold** enhancement due to MnO₂ incorporation. In contrast, the **MnO₂/CNT MEA** delivers a maximum capacitance of 1.8F/cm² or 240F/cm³, more than **200-fold** enhancement due to MnO₂ pseudocapacitance. These results highlight the advantages of using the advanced 3-D nanostructured MnO₂/CNT microelectrode arrays.

Fabrication and characterization of the CNT/MnO₂ planar and MEA ultracapacitors will be presented.

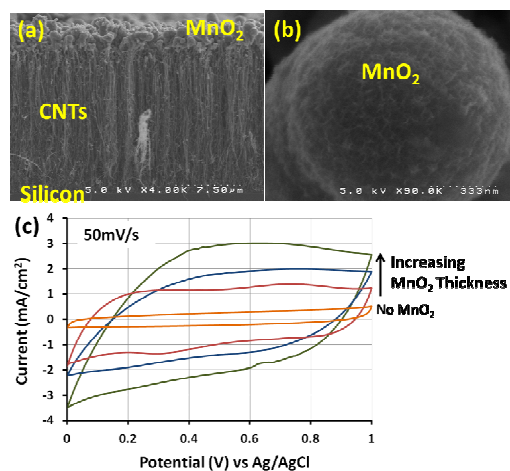


Figure 1. (a) Cross-section SEM of MnO₂ coated CNT planar electrode; (b) High resolution image of porous MnO₂; CVs recorded at 50mV/s showing enhancement in capacitance with addition of MnO₂ on CNTs.

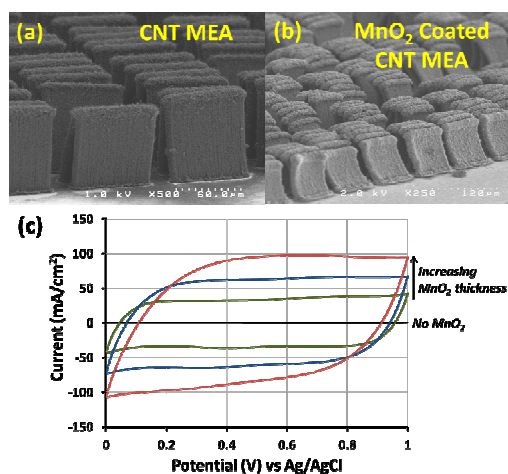


Figure 2. (a) Tilt-view- SEM micrograph of the CNT MEA; (b) Tilt-view- SEM micrograph of the MnO₂ coated CNT MEA; (c) CVs recorded at 50mV/s showing enhancement in capacitance with addition of MnO₂ on CNTs, much greater than the planar ultracapacitor electrode.

References:

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3. G. Lota, K. Lota and E. Frackowiak, *Electrochem. Commun.*, 9, 1828 (2007).